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### **Diversity-Oriented Synthesis of Tamoxifen-type Tetrasubstituted Olefins**

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Tamoxifen is the most important anti-breast cancer drug in clinical use and has the potential to be used as a chemopreventive breast cancer agent.<sup>1</sup> It is a selective estrogen receptor modulator (SERM) with anti-estrogenic properties in breast and estrogenic effects in tissues such as bone and the cardiovascular system. Although these aspects of tamoxifen have stimulated research for a superior and ideal SERM, rapid and systematic screenings have been hampered by the lack of a general and diversity-oriented<sup>2</sup> synthetic scheme for tamoxifen-type tetrasubstituted olefins. In addition, insufficient stereoselectivity, which is often encountered with the widely used dehydration reaction<sup>3</sup> and McMurry coupling,<sup>4</sup> has been a bottleneck for rapid evaluation.<sup>5</sup> As an alternative to these procedures, Miller, Cummins, Knochel, and Armstrong have developed metal-mediated synthetic methods for tamoxifen.<sup>6</sup>

As a part of our program aimed at the development of a general and diversity-oriented synthetic scheme for multisubstituted olefins,<sup>7</sup> we have developed such a synthetic scheme for tamoxifen-type tetrasubstituted olefins utilizing our removable directing 2-pyridyl-silyl group.<sup>8,9</sup> We envisaged that, as an initial step of our campaign, the regio- and stereoselective addition of an arylmetal across 1-butynyldimethyl(2-pyridyl)silane (1) would afford a suitably substituted alkenylmetal compound through the agency of a complex-induced proximity effect of the pyridyl group on silicon. The sequential arylations at C–M and C–Si bonds (Pd-catalyzed cross-coupling reactions) of the resultant pyridylsilyl-substituted alkenylmetal compound would afford the targeted tamoxifen-type tetrasubstituted olefins. Because all aryl groups of this tetrasubstituted olefin structure stem from readily available aryl halides, the synthesis would be sufficiently diversity-oriented.



In early experiments, we found that the addition of Grignard reagents across 1-butynyldimethyl(2-pyridyl)silane (1) does not occur at all.<sup>10</sup> Thus, we searched for an efficient procedure for the initial carbometalation process across 1 (eq 1). After many trials, it was found that the regio- and stereoselective addition of PhMgI (1.0 equiv) to 1 proceeded smoothly in the presence of CuI catalyst to afford 2 in 90% yield. The addition did not occur with an equimolar amount of CuI, which indicates that organocuprate might be the reactive species for this addition. The use of PhMgBr, PhMgCl, or Ph<sub>2</sub>Mg in place of PhMgI resulted in lower addition



efficiency. It was also found that the Cu-catalyzed addition did not occur at all with the corresponding 3-pyridyl, 4-pyridyl, and phenylsilanes, which clearly implicates the strong directing effect (complex-induced proximity effect) of the 2-pyridyl group on silicon.<sup>11</sup>

Table 1. Catalytic One-Pot Diarylation of 1

/	N Si Me <sub>2</sub> 1 (1.0 equiv)	1) Ar <sup>1</sup> MgI (1.5 equiv) CuI (30%), Et <sub>2</sub> O, 0 °C 2) Ar <sup>2</sup> I (1.5 equiv) Pd[P( <i>t</i> -Bu) <sub>3</sub> ] <sub>2</sub> (5%) THF, 40 °C	Ar <sup>1</sup> Et	Ar <sup>2</sup> N Si Me <sub>2</sub> 3
run	Ar <sup>1</sup>	Ar <sup>2</sup>	3	yield ( <i>E/Z</i> )
1	$C_{6}H_{5}\left(\mathbf{a}\right)$	$C_6H_5(\mathbf{a})$	3aa	80% (92/8)
2	$C_{6}H_{5}(a)$	$4-\text{MeOC}_6\text{H}_4$ ( <b>b</b> )	3ab	60% (92/8)
3	$C_{6}H_{5}(a)$	$4-Me_2N(CH_2)_2OC_6H_4(c)$	3ac	55% (88/12)
4	$C_{6}H_{5}(a)$	$4-CF_{3}C_{6}H_{4}(\mathbf{d})$	3ad	75% (95/5)
5	$C_{6}H_{5}(a)$	$4\text{-EtOCOC}_{6}\text{H}_{4}(\mathbf{e})$	3ae	58% (94/6)
6	$C_{6}H_{5}(a)$	$4-ClC_{6}H_{4}(\mathbf{f})$	3af	69% (94/6)
7	$3-ClC_{6}H_{4}(\mathbf{g})$	$4-Me_2N(CH_2)_2OC_6H_4(c)$	3gc	55% (92/8)
8	$3\text{-ClC}_{6}\text{H}_{4}\left(\mathbf{g}\right)$	$4\text{-MeC}_{6}\text{H}_{4}\left(\mathbf{h}\right)$	3gh	79% (92/8)

With the feasibility of an initial carbometalation process, we next examined the one-pot diarylation of **1** through the catalytic carbomagnesation/cross-coupling sequence. After extensive screening of catalyst for the cross-coupling step,  $Pd[P(t-Bu)_3]_2^{12}$  was found to effect the Kumada–Tamao–Corriu-type cross-coupling<sup>13</sup> of the resultant alkenylmagnesium compound and aryl iodide in good overall yield. Thus, by simply varying  $Ar^1MgI$  and  $Ar^2I$ , we could efficiently obtain various structurally and electronically diverse alkenylsilanes **3** (Table 1).<sup>14</sup> The two aryl groups ( $Ar^1$  and  $Ar^2$ ) are introduced in a cis fashion, which is in accordance with syn carbometalation and retention of stereochemistry during the subsequent cross-coupling. The aryl group containing the basic side chain found in tamoxifen can be installed with ease (runs 3 and 7). Interestingly, the ester group also tolerated the conditions of this cross-coupling reaction (run 5).

With the procedure of stereoselective one-pot diarylation of **1** in hand, we investigated the cross-coupling of **3** at the C–Si bond (Hiyama cross-coupling)<sup>7,15</sup> as a final step toward tamoxifen-type tetrasubstituted olefins. However, extensive screening of catalyst and activator to transfer these sterically condensed alkenyl groups

Ar <sup>1</sup> Et	Ar <sup>2</sup> N Si Me <sub>2</sub>	1) BCl <sub>3</sub> (2.2 equiv) CH <sub>2</sub> Cl <sub>2</sub> , -40 °C, 5 h 2) pinacol, Et <sub>3</sub> N <i>one-pot</i>	Ar <sup>1</sup> Et 4
run	<b>3</b> ( <i>E</i> / <i>Z</i> )	4	yield (Z/E)
1	<b>3aa</b> (94/6)	4aa	82% (98/2)
$2^a$	<b>3ac</b> (88/12	2) <b>4ac</b>	65% (94/6)
3	3ad (95/5)	4ad	80% (99/1)
4	<b>3ae</b> (94/6)	4ae	64% (>99/1)
$5^a$	<b>3gc</b> (92/8)	4gc	77% (95/5)
6	<b>3gh</b> (92/8)	4gh	73% (97/3)

<sup>*a*</sup> 3.3 equiv of BCl<sub>3</sub> was employed. After the treatment with pinacol, the mixture was treated with  $Cs_2CO_3$  (10 equiv) in toluene (80 °C, 11 h).

Table 3. Suzuki-Miyaura Coupling of 4 with Aryl lodides

Ar	Ar <sup>2</sup> B(pin) Et	+ Ar <sup>3</sup> -1 (1.2 equiv) Pd[P( <i>t</i> -Bu) <sub>3</sub> ] <sub>2</sub> ( NaOH/H <sub>2</sub> O (3.0 THF 60 °C, 24 h	5%) equiv)	Ar <sup>1</sup> Et 5
run	4 ( <i>Z</i> / <i>E</i> )	Ar <sup>3</sup>	5	yield ( <i>E</i> / <i>Z</i> )
1	<b>4aa</b> (97/3)	$4-Me_2N(CH_2)_2OC_6H_4(c)$	5aac	95% (99/1)
2	4aa (97/3)	$4-MeC_{6}H_{4}(\mathbf{h})$	5aah	96% (99/1)
3	4ac (94/6)	$C_{6}H_{5}\left(\mathbf{a}\right)$	5aca	98% (5/95)
4	<b>4ac</b> (94/6)	$4-\text{MeOC}_6\text{H}_4$ ( <b>b</b> )	5acb	95% (5/95)
5	<b>4ac</b> (94/6)	$3-MeOC_6H_4$ (i)	5aci	92% (95/5)
6	<b>4ad</b> (99/1)	$4-\text{MeOC}_6\text{H}_4$ ( <b>b</b> )	5adb	97% (>99/1)
7	<b>4ad</b> (99/1)	$4-ClC_6H_4(\mathbf{f})$	5adf	90% (>99/1)
8	<b>4ad</b> (99/1)	$2-MeOC_6H_4(\mathbf{j})$	5adj	95% (>99/1)
9	<b>4ad</b> (99/1)	3-pyridyl (k)	5adk	67% (>99/1)
10	<b>4gc</b> (95/5)	$4-MeOC_6H_4$ ( <b>b</b> )	5gcb	80% (4/96)
11	<b>4gc</b> (95/5)	$3-MeC_{6}H_{4}(\mathbf{l})$	5gcl	82% (98/2)
12	4gc (97/3)	3-thienyl (m)	5gcm	87% (99/1)
13	<b>4gh</b> (97/3)	$2-\text{MeC}_6\text{H}_4(\mathbf{n})$	5ghn	93% (>99/1)
14	<b>4gh</b> (99/1)	$3,5-F_2C_6H_3(0)$	5gho	98% (>99/1)
15	<b>4gh</b> (99/1)	1-naphthyl ( <b>p</b> )	5ghp	99% (>99/1)

from silicon met with no success. Thus, we turned our attention to Si/B exchange reaction borodesilylation of **3** to achieve the installation of a third aryl group (Ar<sup>3</sup>) by means of Suzuki–Miyaura coupling at the resulting C–B bond.<sup>16</sup>

The stereoselective borodesilylation<sup>17</sup> of **3** was found to occur with BCl<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -41 °C, and subsequent treatment of this mixture with pinacol and triethylamine (one-pot) afforded **4** in good yield with retention of stereochemistry (Table 2). For compounds bearing the tamoxifen basic side chain (Me<sub>2</sub>NCH<sub>2</sub>CH<sub>2</sub>O-) such as **3ac** and **3gc**, it was necessary to treat the mixture with a base such as Cs<sub>2</sub>CO<sub>3</sub> to remove a boron residue coordinated to the nitrogen. Interestingly, it was found that the isomeric purities of **4** were greater that those of the starting alkenylsilanes in all cases examined. This may be due to the reactivity difference of isomeric alkenylsilanes in borodesilylation.

Finally, the Suzuki–Miyaura coupling of **4** with aryl iodides (Ar<sup>3</sup>I) afforded the targeted tamoxifen-type tetrasubstituted olefin **5**. An extensive optimization of this final step revealed that the catalyst/base/additive combination of Pd[P(*t*-Bu)<sub>3</sub>]<sub>2</sub>/NaOH/H<sub>2</sub>O (5 mol %, 3.0 equiv, 3.0 equiv) effects the desired cross-coupling in extremely high yields in THF at 60 °C (Table 3).<sup>12a</sup> A variety of aryl groups including heteroaryl groups can be installed in the final tetrasubstituted olefin structure. Tamoxifen (**5aca**) and its derivatives can be prepared with ease. It should also be mentioned that this final cross-coupling step also helped to increase the isomeric purities of the final olefins **5** (>99% in most cases) presumably because of the reactivity difference of isomeric **4** in the cross-coupling reaction.

In summary, we have established a general synthetic scheme for tamoxifen-type tetrasubstituted olefins based on the novel Cu-catalyzed carbomagnesation across alkynyl(2-pyridyl)silane. By using this synthetic scheme, we could prepare a wide array of electronically and structurally diverse tetrasubstituted olefins in a regiocontrolled, stereocontrolled, and diversity-oriented manner. Noteworthy features are that (i) the three aryl groups, which are believed to be important (essential) for anti-estrogenic activity, can be varied at will because they all stem from readily available aryl iodides, and (ii) any stereo- and regioisomers can, in principle, be prepared by simply changing the applying order of aryl iodides into the sequence. Although we have focused our investigation on the synthesis of tamoxifen-type olefins in this study, our synthetic scheme should be easily expanded to the construction of a more general tetrasubstituted olefin structure.

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**Supporting Information Available:** Experimental procedures and analytical and spectroscopic data of compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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